

# Alkali Bis(trifluoromethylsulfonyl)imide Molten Salts as New Solvents for Catalysis

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## Low Temperature Molten Salts

### Ionic Liquid

- Organic cation and anion
- $T_m < 100\text{ °C}$
- $T_{\text{stable}} < 250\text{ °C}$  [1]

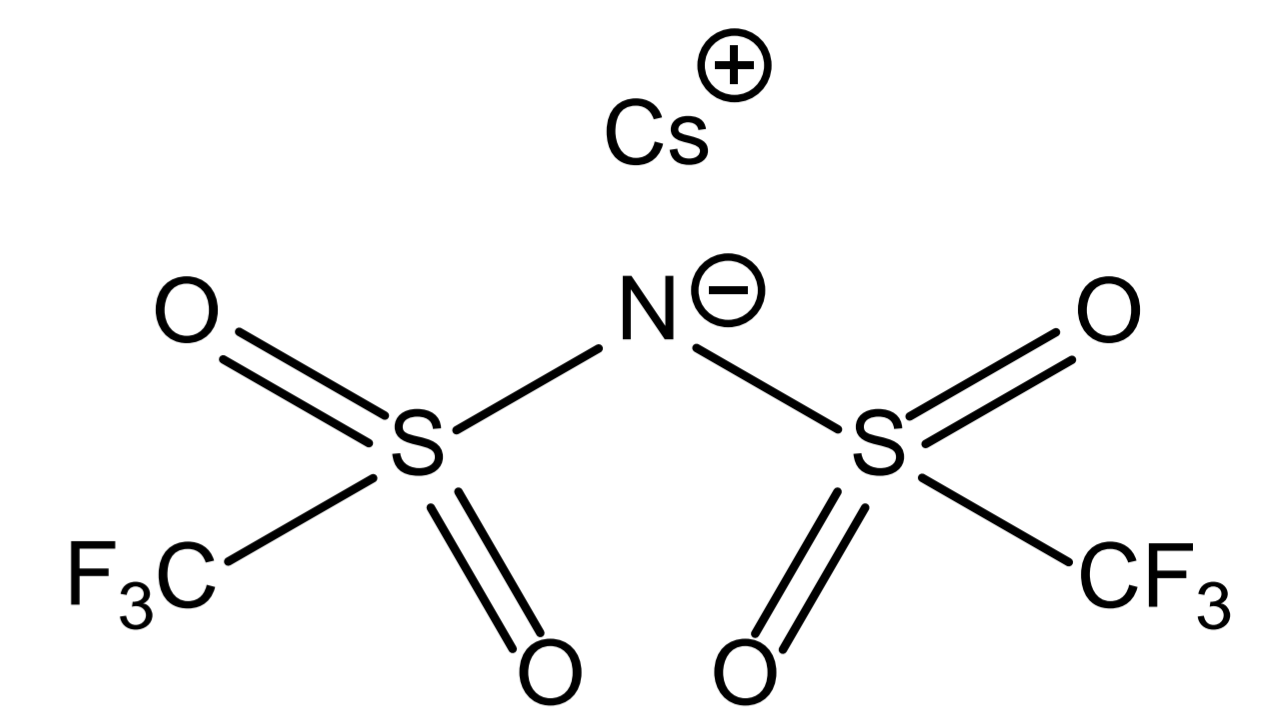
### Low temperature molten salts

- Inorganic cation + organic anion
- Salt mixtures
- $100\text{ °C} < T_m < 250\text{ °C}$
- $T_{\text{stable}} > 250\text{ °C}$

### High temperature molten salts

- Inorganic cation + anion
- $T_m > 400\text{ °C}$
- $T_{\text{stable}} \gg 400\text{ °C}$

- Bridge the gap between ionic liquids and classical high temperature molten salts
- Novel class of reaction media for reactions at temperatures between  $150\text{ °C}$  and  $350\text{ °C}$
- Enhanced reactivity
- Tuneable selectivity
- Basic conditions possible
- Catalyst stabilization



Cesium  
Bis(trifluoromethyl-  
sulfonyl)imide

**CsNTf<sub>2</sub>**

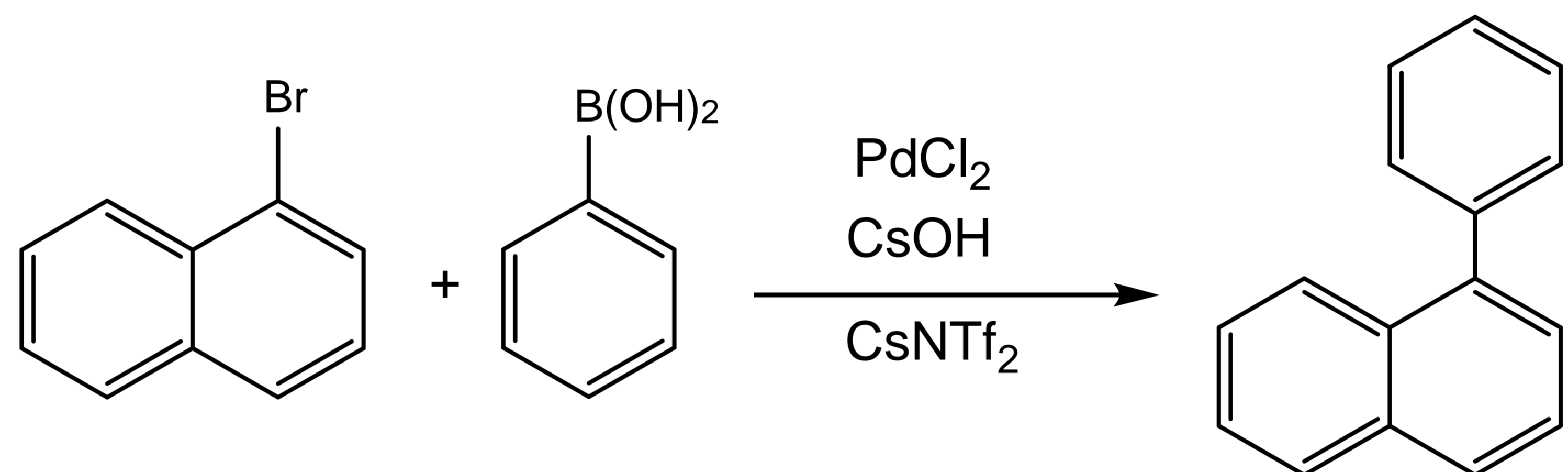
$T_m / \text{°C}$	120
$T_{\text{decomp}} / \text{°C}$	462
$\rho_{200\text{ °C}} / \text{g cm}^{-3}$	2.21
$\eta_{200\text{ °C}} / \text{mPas}$	15

## Experimental Setup

### Test reaction: Suzuki cross-coupling

#### Reaction conditions:

- CsNTf<sub>2</sub> as solvent (2 eq.)
- CsOH as base (1 eq.)
- Inert gas: Argon, atmospheric pressure
- $180\text{ °C}$
- Batch experiments in a glass flask with magnetic stirring bar
- Sample analysis with offline-GC



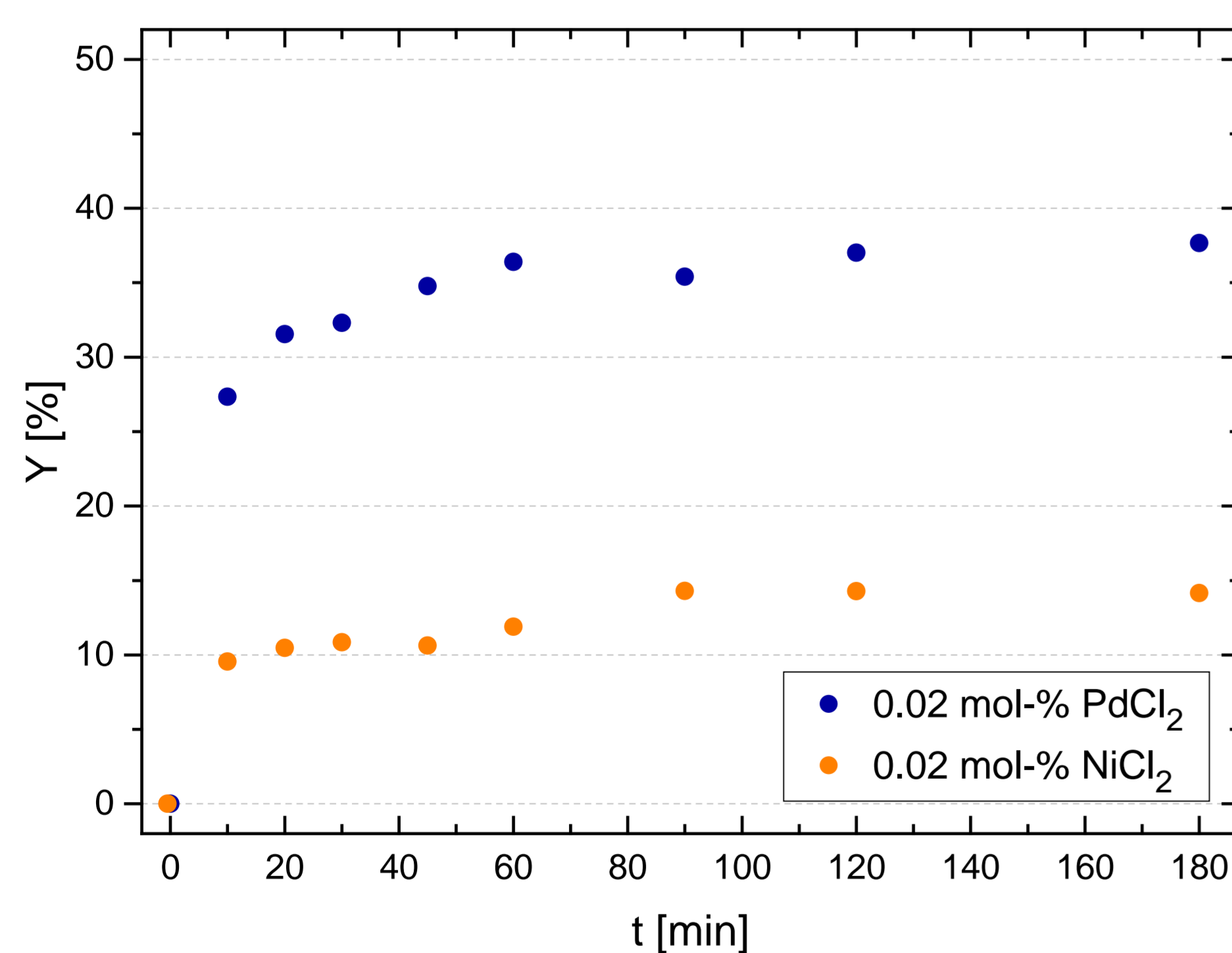
#### Catalyst System:

- PdCl<sub>2</sub> (0.02 mol-%) without further treatment
- Liquid-liquid biphasic system with the catalyst in the molten salt phase

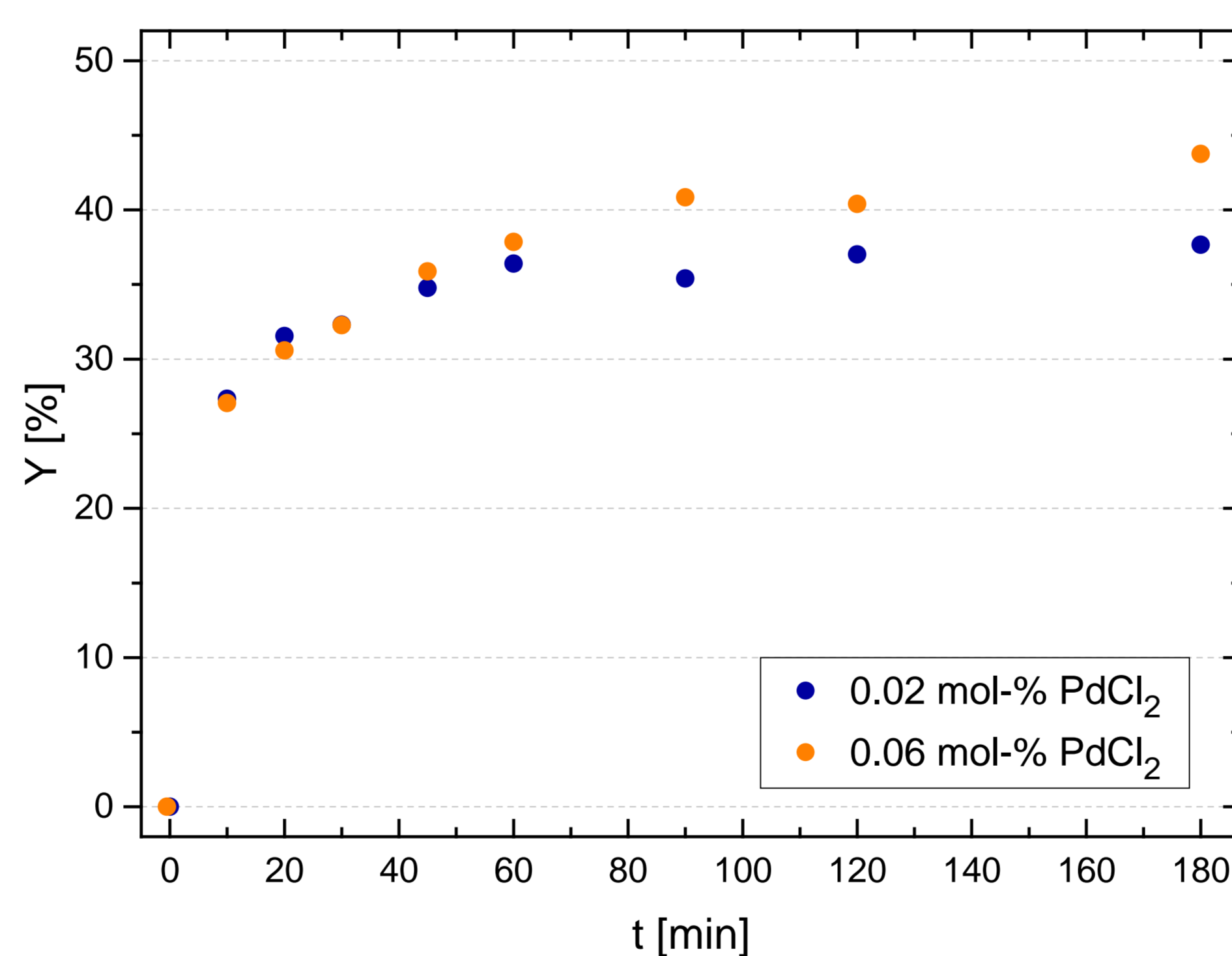
#### Variations:

- Phenylboronic acid derivatives
- Catalyst or Catalyst amount

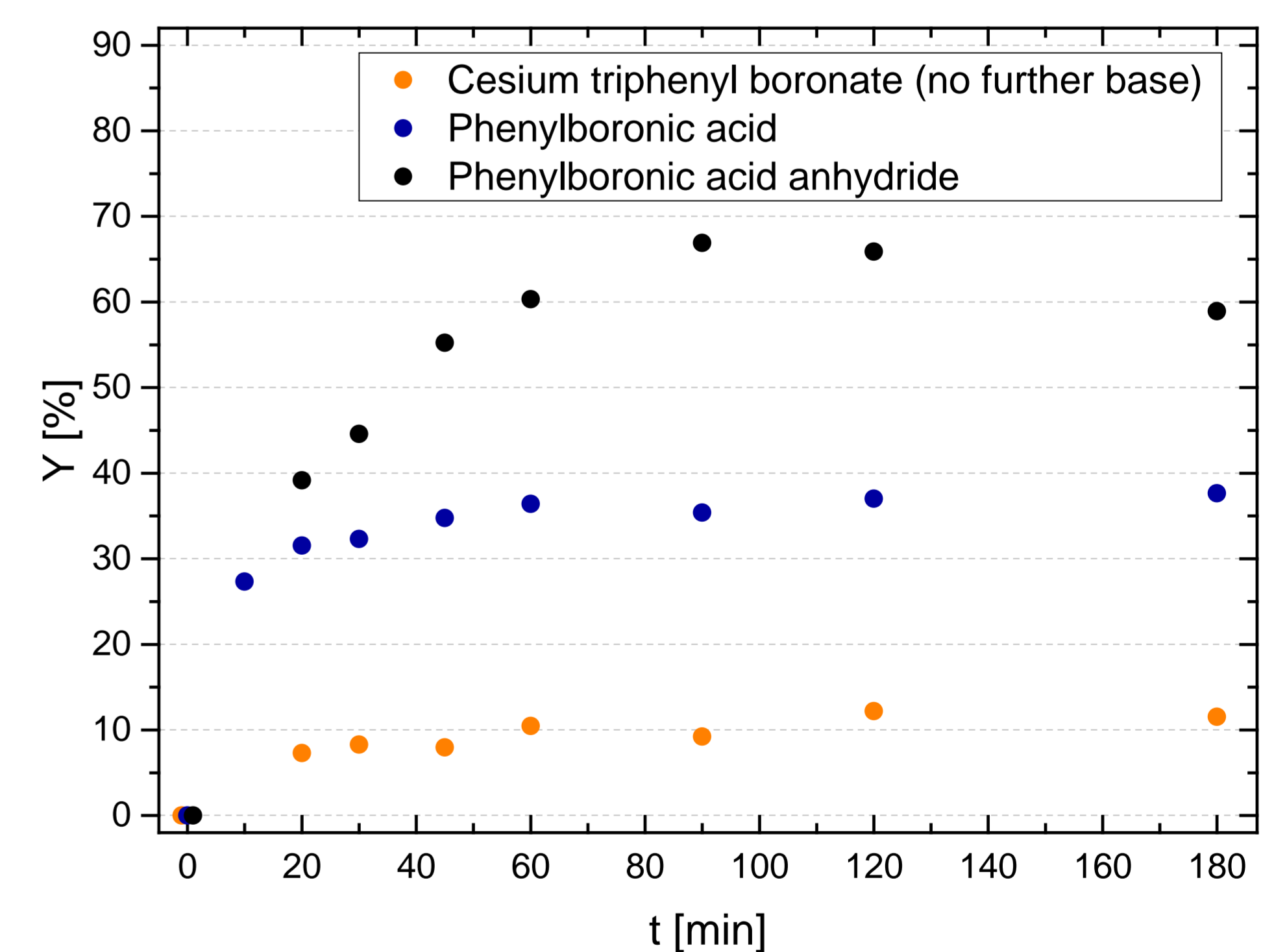
## First Results



Moderate Yields with ligandless Pd-catalyst; even with NiCl<sub>2</sub>



No significant improvement with more catalyst



Best results with Phenylboronic anhydride

## Outlook

- Further experiments with Phenylboronic anhydride
- Temperature variations
- Optimization of the experimental setup (e.g. enlarge of interface area by better stirring)
- Variations of CsNTf<sub>2</sub> amount

[1] S. Werner, M. Haumann, P. Wasserscheid, Annual Review of Chemical and Biomolecular Engineering 1 (2010) 203-230.